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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1901 Excerpt: .chromium alums. With α and β , equal to zero, the solubilities of the vanadium alums are obtained, and by eliminating the second term as well, those of the aluminium salts. We thus have a general solubility equation, all the terms of which can be referred to two variables; the one of these, α , applying to the one class of elements in the compounds, the other, β , to the second class. I believe that aside from the instance of addition properties, which the solubilities certainly are not, this is the first case in which a mathematical relation between the corresponding properties of a class of compounds has been found capable of expression. Below I give the values of α and β for each of the alkali metals, and the solubilities as calculated by the general formula. With the latter, the column D contains the difference between the calculated and observed solubilities, and D the variation from the values serving as a basis for the derivation of the constants. - p Cs. 2 6 5 10.945 Eb. 27 11 19 12.254 Tl. 57 56 4 20.5406 NH. 73 45 5 38.943 In the following the solubilities are expressed in hundredths of a gram-molecule: In conclusion I may state that while the data now at hand are too few for accurate analysis, such determinations as I have made at temperatures other than 25 indicate that a general solubility formula for all temperatures will not be difficult of derivation. I...



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